

Facile Lithiation of Co-ordinated *t*-Methylphosphines in Bis-(2,6-dimethoxyphenyl)nickel(II) Complexes

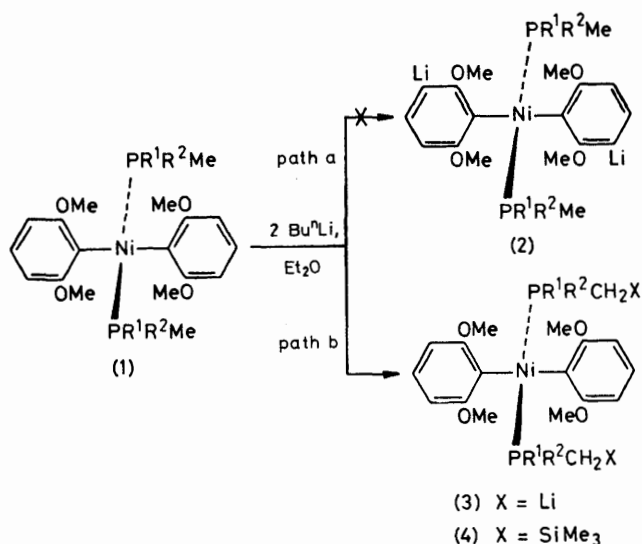
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Summary Reactions of *n*-butyl-lithium with the bis-(2,6-dimethoxyphenyl)nickel(II) complexes, *trans*-[Ni{C₆H₃(OMe)₂-2,6}₂(PR₃)₂] (PR₃ = PMe₃ and PMe₂Ph), in diethyl ether resulted in the facile lithiation at the methyl carbon in the PR₃ ligands.

THE bis-(2,6-dimethoxyphenyl)nickel(II) complexes, *trans*-[Ni{C₆H₃(OMe)₂-2,6}₂(PR₃)₂] [PR₃ = PMe₃ (**1a**) and PMe₂Ph (**1b**)], are easily obtainable, highly crystalline, and air-stable.¹ Since methoxyphenyl derivatives are well known to be lithiated by *n*-butyl-lithium *ortho* to the methoxy-group,² we attempted the reaction for complexes (**1a**) and (**1b**) in order to obtain the 3-lithiated intermediates (**2**) (path a, Scheme). We report herein the unexpected result that a methyl proton in the PR₃ ligands was substituted easily by lithium (**3**) (path b, Scheme). No report of unambiguous proton abstraction from a co-ordinated *t*-phosphine has yet appeared,³ although proton abstractions from free *t*-phosphines⁴ and co-ordinated diphosphines⁵ have been discussed.

Treatment of compound (**1a**) with an excess of *n*-butyl-lithium in dry diethyl ether at 0 °C under nitrogen gave, in less than a few minutes, a light yellow precipitate. This was stirred at room temperature for 2 h and then quenched with D₂O. The ¹H n.m.r. spectrum (CDCl₃) of deuteriated (**1a**) was essentially identical with that of (**1a**), but showed a decrease in the intensity of the PMe₃ proton resonances by *ca.* two protons. When the lithiated intermediate was treated with Me₃SiCl, an air-stable orange-yellow crystalline complex, *trans*-[Ni{C₆H₃(OMe)₂-2,6}₂(PMe₂CH₂SiMe₃)₂] (**4a**), m.p. 87–88 °C (45–55% yield), was obtained [¹H n.m.r. (CDCl₃) δ -0.07 (s, 18 H, SiMe), 0.67 (t, 4 H, J_F 10 Hz, PCH₂Si), 0.79 (t, 12 H, J_F 7.5 Hz, PMe), 3.86 (s, 12 H, OMe),



a; R¹ = R² = Me
b; R¹ = Me, R² = Ph

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6.29 (d, 4 H, J_H 7.5 Hz, *m*-H), and 6.91 (br t, 2 H, J_H 7.5 Hz, *p*-H)]. These observations suggest that a proton was abstracted by *n*-butyl-lithium from PMe₃ ligands rather than from the 2,6-dimethoxyphenyl groups. Compound (**4a**) was treated with carbon monoxide (1 atmosphere) in the presence of MeI and decomposed, in less than 2 h at room temperature, to give [C₆H₃(OMe)₂-2,6]₂CO and the [Me₃PCH₂SiMe₃]⁺ salt.

Reaction of compound (**1b**) with n-butyl-lithium followed by treatment with Me_3SiCl gave an analogous complex, *trans*- $[\text{Ni}\{\text{C}_6\text{H}_3(\text{OMe})_2-2,6\}_2(\text{PMePhCH}_2\text{SiMe}_3)_2]$ (**4b**) (70–80%), but as a mixture of the *racemo*- and *meso*-isomers with respect to the two asymmetric phosphine ligands [^1H n.m.r. (CDCl_3) δ –0.30 (s, SiMe), 0.96 (overlapped t, J_{P} 11 Hz, PCH_2Si), 1.04 (overlapped t, J_{P} 7 Hz, PMe), 3.47 (s, OMe, *meso*), 3.58 (s, OMe, *racemo*), 3.69 (s, OMe, *meso*), 6.03 (d, J_{H} 8, *m*-H, *meso*), 6.08 (d, J_{H} 8, *m*-H, *racemo*), 6.14 (d, J_{H} 8, *m*-H, *meso*), and 6.7–7.2 (m, ArH)]. The relative *meso*:*racemo* ratio was 62:38 for the initial product and varied in the recrystallized fractions.

Two other complexes, *trans*- $[\text{Ni}\{\text{C}_6\text{H}_3(\text{OMe})_2-2,6\}(\text{C}_6\text{H}_2\text{Me}_3-2,4,6)(\text{PMe}_3)_2]$ and *trans*- $[\text{Ni}\{\text{C}_6\text{H}_2\text{Me}_3-2,4,6\}(\text{C}_6\text{H}_4\text{Me}-2)(\text{PMe}_3)_2]$, reacted slowly with n-butyl-lithium even in the presence of tetramethylethylenediamine (15–22 h), and gave, on treatment with Me_3SiCl , *trans*- $[\text{Ni}\{\text{C}_6\text{H}_3(\text{OMe})_2-$

$2,6\}(\text{C}_6\text{H}_2\text{Me}_3-2,4,6)(\text{PMe}_3)(\text{PMe}_2\text{CH}_2\text{SiMe}_3)]$ (70%) and *trans*- $[\text{Ni}\{\text{C}_6\text{H}_2\text{Me}_3-2,4,6\}(\text{C}_6\text{H}_4\text{Me}-2)(\text{PMe}_2\text{CH}_2\text{SiMe}_3)_2]$ (30%), respectively.

These results clearly indicate that proton abstraction by n-butyl-lithium from t-methylphosphine co-ordinated to a transition metal has occurred. The high reactivity of bis-(2,6-dimethoxyphenyl)nickel(II) complexes is probably due to the presence of four methoxy-groups, which increase the reactivity of n-butyl-lithium by co-ordination and/or which stabilize the lithiated intermediate by forming bicyclic rings with the *COO*-chelated lithium ion, both above and below the nickel(II) co-ordination plane. The difficult double-lithiation of *trans*- $[\text{Ni}\{\text{C}_6\text{H}_3(\text{OMe})_2-2,6\}(\text{C}_6\text{H}_2\text{Me}_3-2,4,6)(\text{PMe}_3)_2]$ is tentatively attributed to steric interference by the two bulky aryl groups.

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